Dynamics of the Atmosphere and the Ocean

Lecture 1

Szymon Malinowski

2025-2026 Fall



Acknowledgments to G.K. Vallis, A lot of material in this lecture comes from: Atmospheric and Oceanic Fluid Dynamics. Available from www.princeton.edu/gkv/aofd (and later book published by Cambridge University Press)

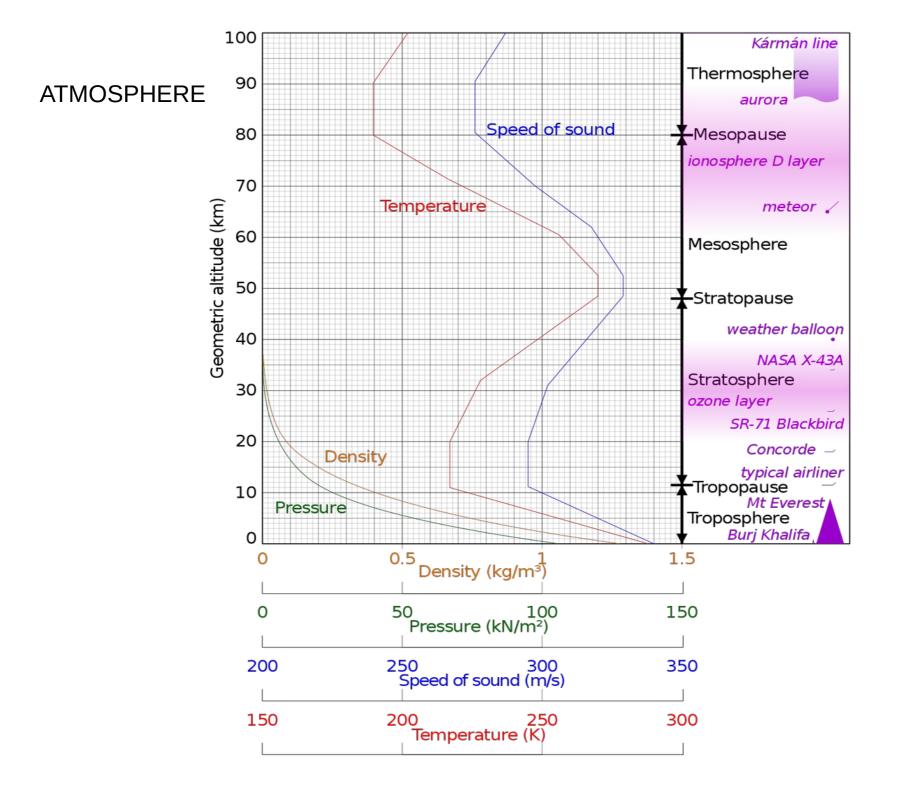
James Holton's "An introduction to dynamic meteororlogy" is the must read textbook and should be a companion to this course



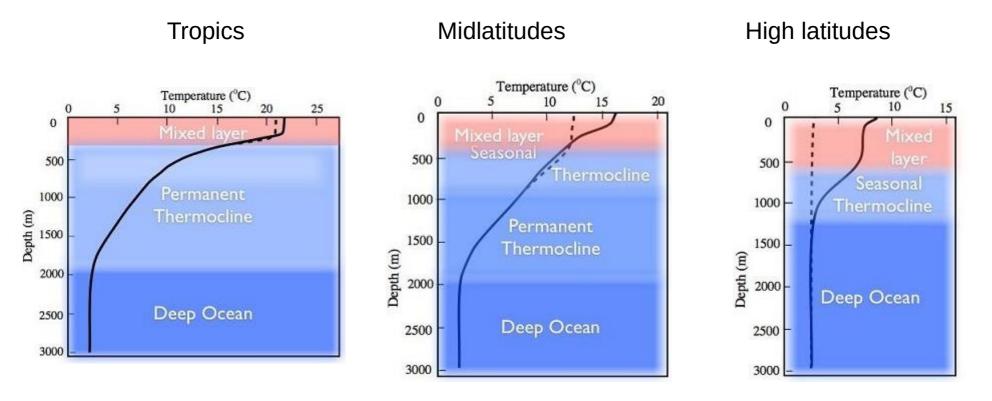
Geophysical fluid dynamics, in its broadest meaning, refers to the fluid dynamics of naturally occurring flows, such as lava flows, oceans, and planetary atmospheres, on Earth and other planets.

We will focus on the applications of geophysical fluid dynamics to the Earth's atmosphere and ocean.





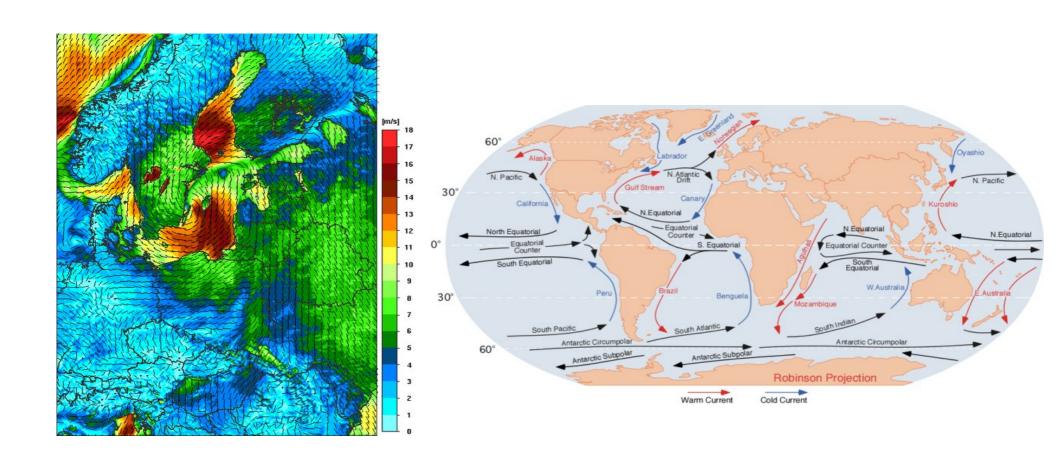
**OCEAN** 



Not only temperature, but salinity as well

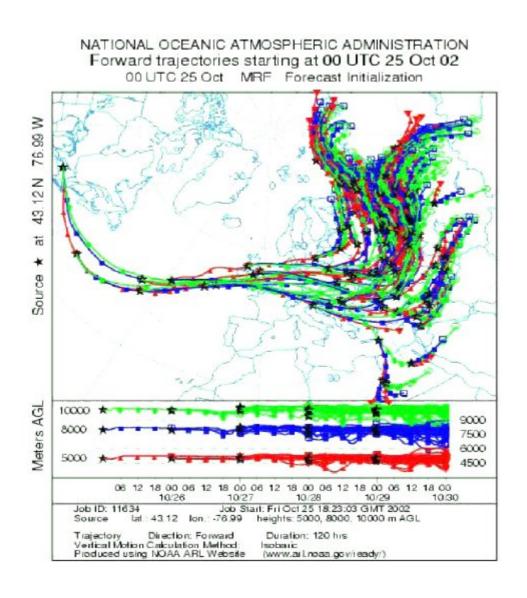
### Eulerian specification of the flow.

The **Eulerian** specification of the flow field is a way of looking at fluid motion that focuses on specific locations in the space (X) through which the fluid flows as time (t) passes. Velocity V(X,t) characterizes the flow.



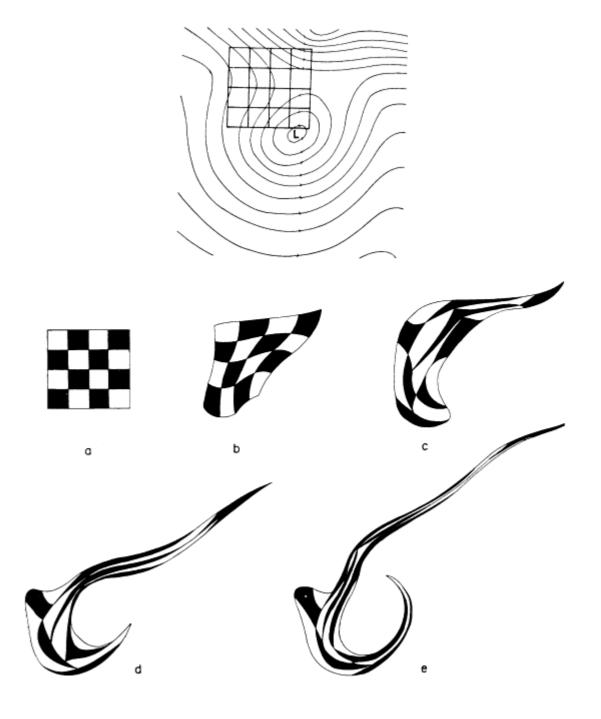
### Lagrangian specification of the flow

The Lagrangian specification of the flow field is a way of looking at fluid motion where the observer follows an individual fluid parcel as it moves through space and time. Position of the parcel is X(t) and properties evolve with time.



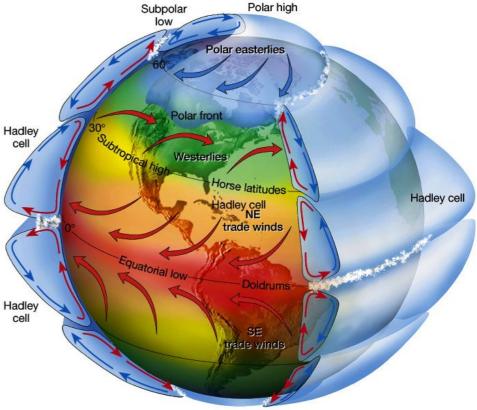
Practical problem:

deformation of Lagrangian parcel in time...

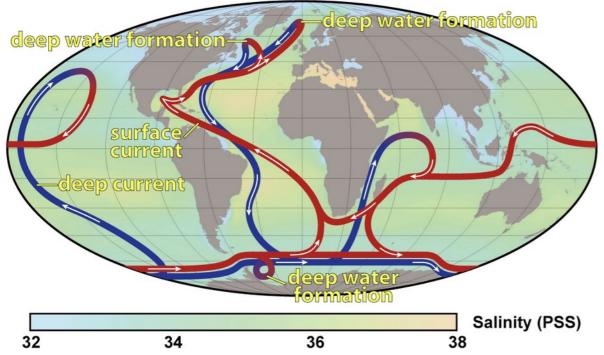


R.A. Pielke, M. Uliasz, Use of meteorological models as input to regional and mesoscale air quality models — limitations and strengths, Atmospheric Environment, Volume 32, Issue 8, 1 April 1998, Pages 1455-1466, ISSN 1352-2310, 10.1016/S1352-2310(97)00140-4.

## Global-scale circulations



### **Thermohaline Circulation**



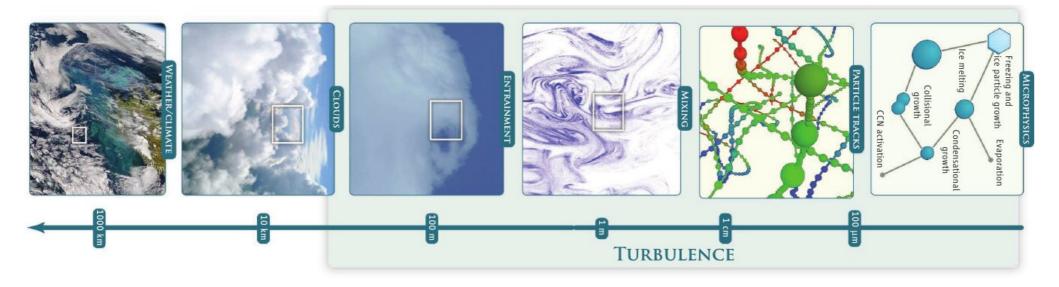
Both oceanic and atmospheric flows span over a wide range of scales: from global scale down to viscosity scale.

Flow type	Scale (m)	Scale name
	10 <sup>-7</sup> m (0.1 μm) or less	mean free path
viscous flow	10 <sup>-3</sup> m (1mm)	
dissipation-scale eddies	10 <sup>-2</sup> m (1cm)	turbulence
small eddies	10 <sup>-1</sup> m (10cm)	
dust devils, eddies, voritces, surface waves	1-10m	
wind blows, surface waves	10-100m	
tornadoes,	100-1000m	
convective clouds, boundary layer eddies	10 <sup>3</sup> -10 <sup>4</sup> m (1-10km)	
mesoscale convective systems, fronts, sea current loops	10 <sup>4</sup> -10 <sup>6</sup> m (10-1000km)	mesoscale
hurricanes,	10 <sup>5</sup> -10 <sup>6</sup> m (100-1000km)	synoptic scale
low and high pressure systems	10 <sup>6</sup> m (1000km)	
clobal circulation	10 <sup>7</sup> m (10000km)	global scale

# Can We Understand Clouds Without Turbulence?

E. Bodenschatz, 1.2 S. P. Malinowski, 3 R. A. Shaw, 4 F. Stratmann 5

Advances at the interface between atmospheric and turbulence research are helping to elucidate fundamental properties of clouds.



### **Material and Eulerian Derivatives**

The material derivative of a scalar  $(\phi)$  and a vector (b) field are given by:

$$\frac{\mathrm{D}\boldsymbol{\phi}}{\mathrm{D}t} = \frac{\partial \boldsymbol{\phi}}{\partial t} + \boldsymbol{v} \cdot \nabla \boldsymbol{\phi}, \qquad \qquad \frac{\mathrm{D}\boldsymbol{b}}{\mathrm{D}t} = \frac{\partial \boldsymbol{b}}{\partial t} + (\boldsymbol{v} \cdot \nabla)\boldsymbol{b}. \tag{D.1}$$

Various material derivatives of integrals are:

$$\frac{\mathrm{D}}{\mathrm{D}t} \int_{V} \phi \, \mathrm{d}V = \int_{V} \left( \frac{\mathrm{D}\phi}{\mathrm{D}t} + \phi \nabla \cdot \boldsymbol{v} \right) \mathrm{d}V = \int_{V} \left( \frac{\partial \phi}{\partial t} + \nabla \cdot (\phi \boldsymbol{v}) \right) \mathrm{d}V, \quad (D.2)$$

$$\frac{\mathrm{D}}{\mathrm{D}t} \int_{V} \mathrm{d}V = \int_{V} \nabla \cdot \mathbf{v} \, \mathrm{d}V, \tag{D.3}$$

$$\frac{\mathrm{D}}{\mathrm{D}t} \int_{V} \rho \phi \, \mathrm{d}V = \int_{V} \rho \frac{\mathrm{D}\phi}{\mathrm{D}t} \, \mathrm{d}V. \tag{D.4}$$

These formulae also hold if  $\phi$  is a vector. The Eulerian derivative of an integral is:

$$\frac{\mathrm{d}}{\mathrm{d}t} \int_{V} \phi \, \mathrm{d}V = \int_{V} \frac{\partial \phi}{\partial t} \, \mathrm{d}V, \tag{D.5}$$

so that

$$\frac{\mathrm{d}}{\mathrm{d}t} \int_{V} \mathrm{d}V = 0 \quad \text{and} \quad \frac{\mathrm{d}}{\mathrm{d}t} \int_{V} \rho \phi \, \mathrm{d}V = \int_{V} \frac{\partial \rho \phi}{\partial t} \, \mathrm{d}V. \tag{D.6}$$

### **Governing Equations**

### **Mass continuity** in Eulerian approach:

consider the flow of mass through a control volume

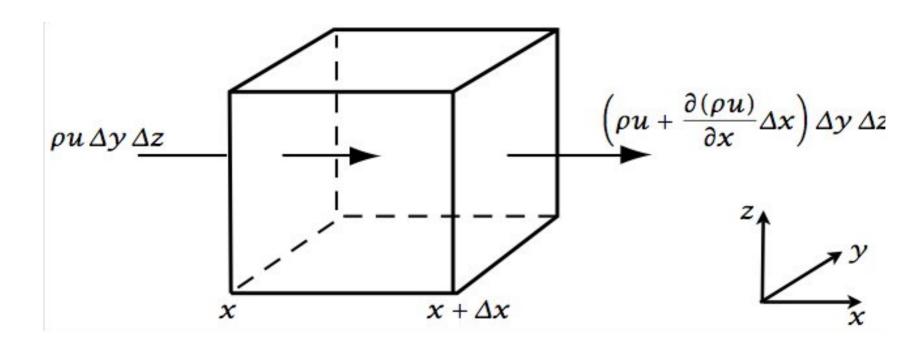


Fig. 1.1 Mass conservation in an Eulerian cuboid control volume.

$$\delta x \delta y \delta z \left[ \frac{\partial \rho}{\partial t} + \frac{\partial (\rho u)}{\partial x} + \frac{\partial (\rho v)}{\partial v} + \frac{\partial (\rho w)}{\partial z} \right] = 0.$$

$$\delta x \delta y \delta z \left[ \frac{\partial \rho}{\partial t} + \frac{\partial (\rho u)}{\partial x} + \frac{\partial (\rho v)}{\partial y} + \frac{\partial (\rho w)}{\partial z} \right] = 0.$$



This [in brackets] equals ZERO, hence....

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{v}) = 0.$$

**Continuity equation = conservation of mass**, which in classical mechanics is strictly fulfilled.

There are alternative derivations, consult Vallis book!!!

One example is by material derivative:

$$\frac{D}{Dt}(\rho\Delta V) = 0$$
where the conserved conserved is the conserved of th

Similar conservation laws are valid for any PASSIVE SCALAR carried by the flow:

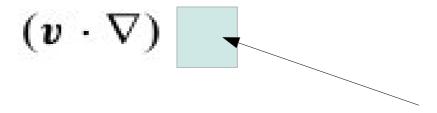
$$\frac{\mathrm{D}\xi}{\mathrm{D}t} = \dot{\xi}.$$

Any scalar with no sources or sinks (r.h.s. =0) is conserved!

E.g. temperature with no heat sources or sinks, dye....

THE ABOVE FORM OF EQUATION IS CRUCIALLY IMPORTANT FOR ALL CONSERVATION LAWS IN FLUID DYNAMICS!

Notice in the l.h.s. the advection operator



This stays for anything!!!

### **Conservation of momentum:**

 $\mathbf{m}(x,y,z,t)$  - the momentum-density field (momentum per unit volume),  $\mathbf{m}=\rho~\mathbf{v}$ 

The total momentum of a volume of fluid is given by the volume integral.

The rate of change of a momentum is given by the material derivative, and by Newton's second law is equal to the force acting on it:

$$\begin{split} \frac{\mathrm{D}}{\mathrm{D}t} \int_{V} \rho \mathbf{v} \, \mathrm{d}V &= \int_{V} \mathbf{F} \, \mathrm{d}V \\ \frac{\mathrm{D}}{\mathrm{D}t} \int_{V} \rho \mathbf{v} \, \mathrm{d}V &= \int_{V} \rho \frac{\mathrm{D}\mathbf{v}}{\mathrm{D}t} \, \mathbf{d}V, \quad \text{acceleration of fluid of density } \rho, \\ \int_{V} \left( \rho \frac{\mathrm{D}\mathbf{v}}{\mathrm{D}t} - \mathbf{F} \right) \, \mathrm{d}V &= 0. \end{split}$$

$$\rho \frac{\mathrm{D} \boldsymbol{v}}{\mathrm{D} t} = \boldsymbol{F}$$

NOTICE THE FORM OF THE ABOVE EQUATION!

NO FORCE = CONSERVATION OF MOMENTUM...

The most common form of **momentum equation**:

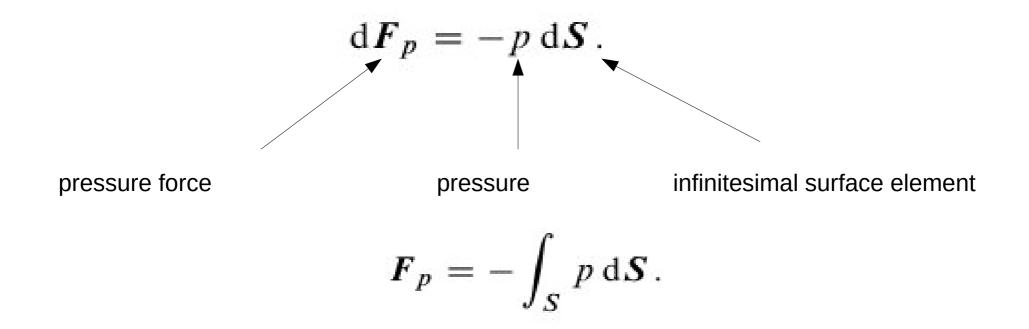
$$\frac{\partial \mathbf{v}}{\partial t} + (\mathbf{v} \cdot \nabla)\mathbf{v} = \frac{\mathbf{F}}{\rho}$$

In the r.h.s. F stays for sum of ALL forces!

Which forces ???

### 1. Pressure force:

at the boundary of a fluid the pressure is the normal force per unit area



with application of divergence theorem one gets:

$$\boldsymbol{F}_p = -\int_V \nabla p \, \mathrm{d}V$$

# 2. Viscosity force:

Many textbooks show that, for most Newtonian fluids, the viscous force per unit volume is equal to  $\mu\Delta v$ , where  $\mu$  is a coefficient of diffusivity. This is an extremely good approximation for most liquids and gases. With this term, the momentum equation becomes:

$$\frac{\partial \mathbf{v}}{\partial t} + (\mathbf{v} \cdot \nabla)\mathbf{v} = -\frac{1}{\rho} \nabla p + \nu \nabla^2 \mathbf{v}$$

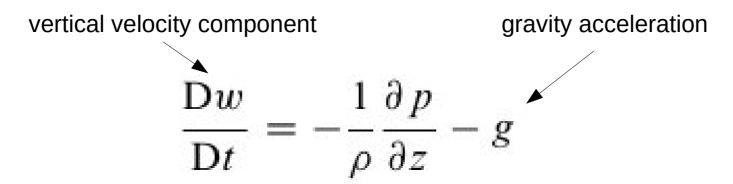
kinematic viscosity,  $v=\mu/\rho$ ,  $v\sim$  (mean free path)\*(mean molecular velocity)

	$\mu \; (\mathrm{kg}\mathrm{m}^{-1}\mathrm{s}^{-1})$	$v (m^2 s^{-1})$
Air	$1.8 \ 10^{-5}$	$1.5 \ 10^{-5}$
Water	$1.1 \ 10^{-3}$	$1.1 \ 10^{-6}$
Mercury	$1.6 \ 10^{-3}$	$1.2 \ 10^{-7}$

## 3. Gravity force.

Let's consider situation with negligible viscosity and important gravity.

Let's focus on a vertical component of the momentum equation (along gravity acceleration).

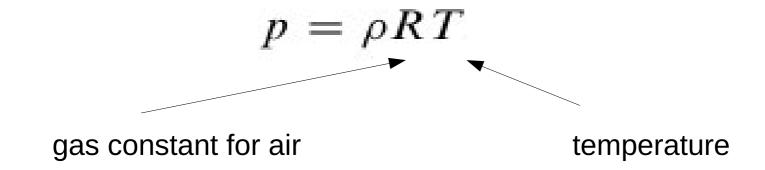


in static situation (no motion) vertical pressure gradient is balanced by gravity acceleration, this is condition for **hydrostatic balance**:

$$\frac{\partial p}{\partial z} = -\rho g$$

Consider, that in many fluids density and pressure can be related. In general, equation which allows to calculate density with use of other properties of fluid is called a **CONSTITUTIVE EQUATION** (in a narrow sense), or EQUATION OF STATE

E.g. for air ideal gas equation can be used as a goos approximation of equation of state:

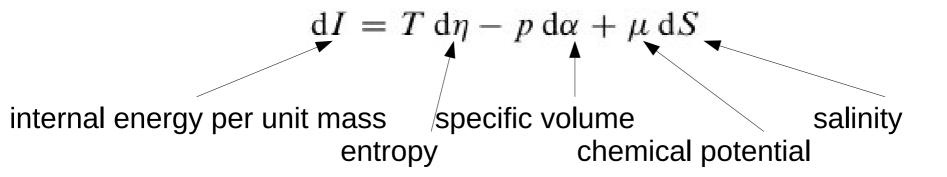


For pure water we usually use:

$$\rho = \rho_0[1-\beta_T(T-T_0)]$$
 thermal expansion coefficient —

For ocean water relation is more complicated, salinity and pressure should be accounted for.

**Energy equation** or the first principle of thermodynamics:



In fact this is conservation equation for internal energy and can be expressed in the form similar to these discussed earlier.

E.g. assuming that air is the ideal gas we can write:

$$\mathrm{d}Q = \mathrm{d}I + p\,\mathrm{d}\alpha$$
 
$$\mathrm{d}Q = c_p\,\mathrm{d}T + p\,\mathrm{d}\alpha \qquad \qquad \mathrm{d}Q = c_p\,\mathrm{d}T - \alpha\,\mathrm{d}p$$
 
$$\alpha = RT/p \text{ and and } c_p - c_v = R$$

Let's consider the material derivative of internal energy:

$$\frac{\mathrm{D}I}{\mathrm{D}t} + p\frac{\mathrm{D}\alpha}{\mathrm{D}t} = \dot{Q}.$$

$$c_v \frac{\mathrm{D}T}{\mathrm{D}t} + p \frac{\mathrm{D}\alpha}{\mathrm{D}t} = \dot{Q}, \quad \text{or} \quad c_p \frac{\mathrm{D}T}{\mathrm{D}t} - \frac{RT}{p} \frac{\mathrm{D}p}{\mathrm{D}t} = \dot{Q}.$$

Finally we get **energy equation** in form:

$$c_v \frac{\mathrm{D}T}{\mathrm{D}t} + p\alpha \nabla \cdot \mathbf{v} = \dot{Q}.$$

# Adiabatic processes:

In adiabatic processes there are no sources and sinks of heat:

$$c_p dT = \alpha dp$$

The **potential temperature**,  $\Theta$  is the temperature that a fluid would have if moved adiabatically to some reference pressure (often 1000 hPa, close to the pressure at the earth's surface). In adiabatic flow the potential temperature of a fluid parcel is conserved, by definition:

$$\frac{\mathrm{D}\theta}{\mathrm{D}t} = 0$$

For an ideal gas:

$$d\eta = c_p d \ln T - R d \ln p$$

$$c_p d \ln \theta = c_p d \ln T - R d \ln p$$

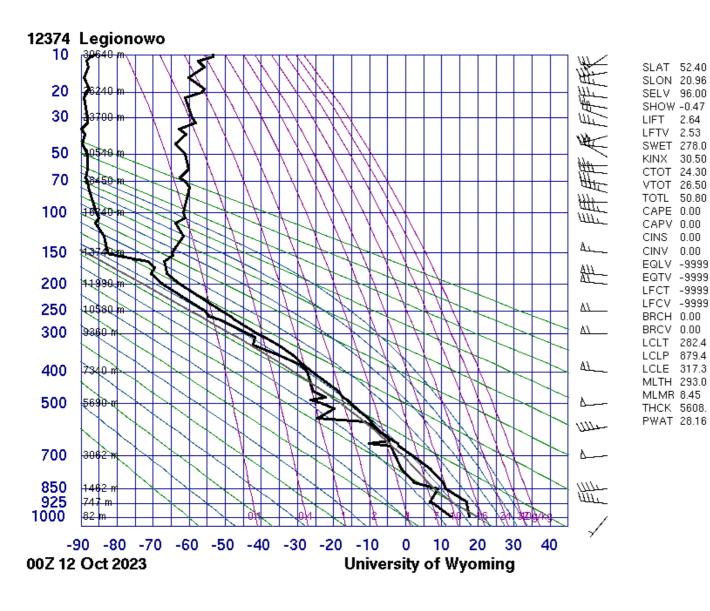
Consequently:

reference pressure 
$$\kappa = R/c_p$$

$$\theta = T\left(\frac{p_R}{p}\right)^{\kappa}$$

Finally we may write a compact form of the energy equation:

$$c_p \frac{\mathrm{D}\theta}{\mathrm{D}t} = \frac{\theta}{T} \dot{Q}$$

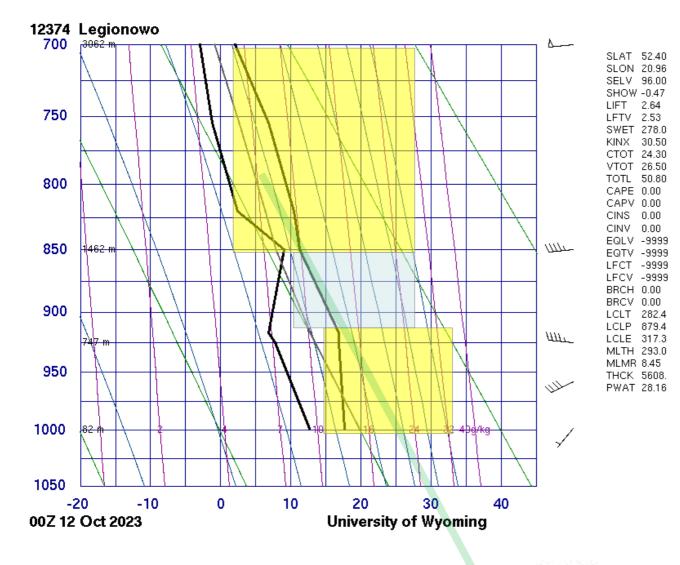


Potential temperature: an application

Vertical profile of the atmosphere over Warsaw from ballon measurement.

Black lines: measured temperature (right) and dew point temperature – right.

Adiabats ( $\theta$ =const) - green lines.



Potential temperature: an application

Vertical profile of the atmosphere over Warsaw from ballon measurement.

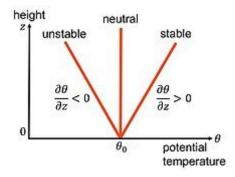
Black lines: measured temperature (right) and dew point temperature – right.

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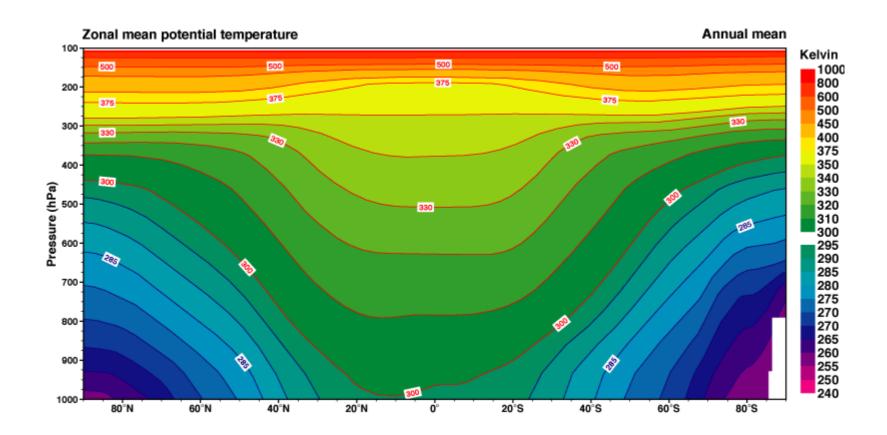
almost neutral

stable

$$\frac{\mathrm{D}\theta}{\mathrm{D}t} = 0$$



# Atmospheric stability and meridional potential temperature distribution



# Forms of the Thermodynamic Equation

# General form

For a parcel of constant composition the thermodynamic equation is

$$T\frac{\mathrm{D}\eta}{\mathrm{D}t} = \dot{Q}$$
 or  $c_p \frac{\mathrm{D}\ln\theta}{\mathrm{D}t} = \frac{1}{T}\dot{Q}$  (T.1)

where  $\eta$  is the entropy,  $\theta$  is the potential temperature,  $c_p \ln \theta = \eta$  and  $\dot{Q}$  is the heating rate. Applying the first law of thermodynamics  $T d\eta = dI + p d\alpha$  gives:

$$\frac{\mathrm{D}I}{\mathrm{D}t} + p \frac{\mathrm{D}\alpha}{\mathrm{D}t} = \dot{Q} \qquad \text{or} \qquad \frac{\mathrm{D}I}{\mathrm{D}t} + RT\nabla \cdot \mathbf{v} = \dot{Q} \tag{T.2}$$

where I is the internal energy.

### The Equations of Motion of a Fluid

For dry air, or for a salt-free liquid, the complete set of equations of motion may be written as follows:

The mass continuity equation:

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{v}) = 0. \tag{EOM.1}$$

If density is constant this reduces to  $\nabla \cdot \mathbf{v} = 0$ .

The momentum equation:

$$\frac{\mathbf{D}\boldsymbol{v}}{\mathbf{D}t} = -\frac{\nabla p}{\rho} + \nu \nabla^2 \boldsymbol{v} + \boldsymbol{F}, \qquad (EOM.2)$$

The thermodynamic equation:

$$\frac{\mathrm{D}\theta}{\mathrm{D}t} = \frac{1}{c_p} \left(\frac{\theta}{T}\right) \dot{Q}. \tag{EOM.3}$$

where  $\dot{Q}$  represents external heating and diffusion, the latter being  $\kappa \nabla^2 \theta$  where  $\kappa$  is the diffusivity.

The equation of state:

$$\rho = g(\theta, p) \tag{EOM.4}$$

where g is a given function. For example, for an ideal gas,  $\rho = p_R^{\kappa}/(R\theta p^{\kappa-1})$ .

## The real-scienific-life example:

1726

### JOURNAL OF THE ATMOSPHERIC SCIENCES

#### VOLUME 6

### Numerical Simulation of Cloud-Clear Air Interfacial Mixing

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(Manuscript received 25 October 2002, in final form 2 February 2004)

$$D/Dt \equiv \partial/\partial t + \mathbf{v} \cdot \nabla$$

$$B \equiv g \left[ \frac{T - T_0}{T_0} + \varepsilon (q_v - q_{v_0}) - q_c \right], \tag{2}$$

$$\frac{D\mathbf{v}}{Dt} = -\nabla \pi + \mathbf{k}B + \nu \nabla^2 \mathbf{v}, \tag{1a}$$

$$\nabla \cdot \mathbf{v} = 0, \tag{1b}$$

$$\frac{DT}{Dt} = \frac{L}{c_p} C_d + \mu_T \nabla^2 T, \tag{1c}$$

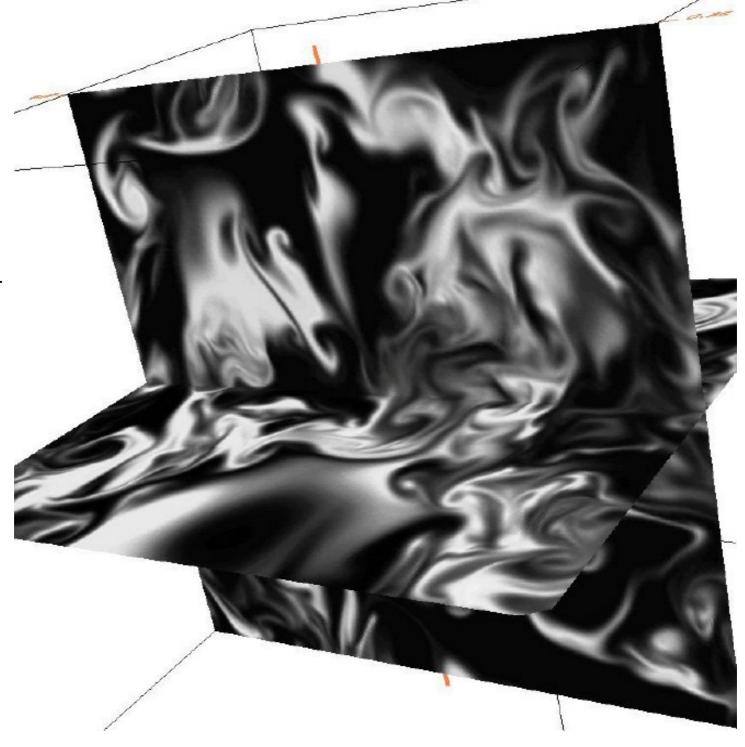
$$\frac{Dq_v}{Dt} = -C_d + \mu_v \nabla^2 q_v, \tag{1d}$$

$$\frac{Dq_c}{Dt} = C_d.$$

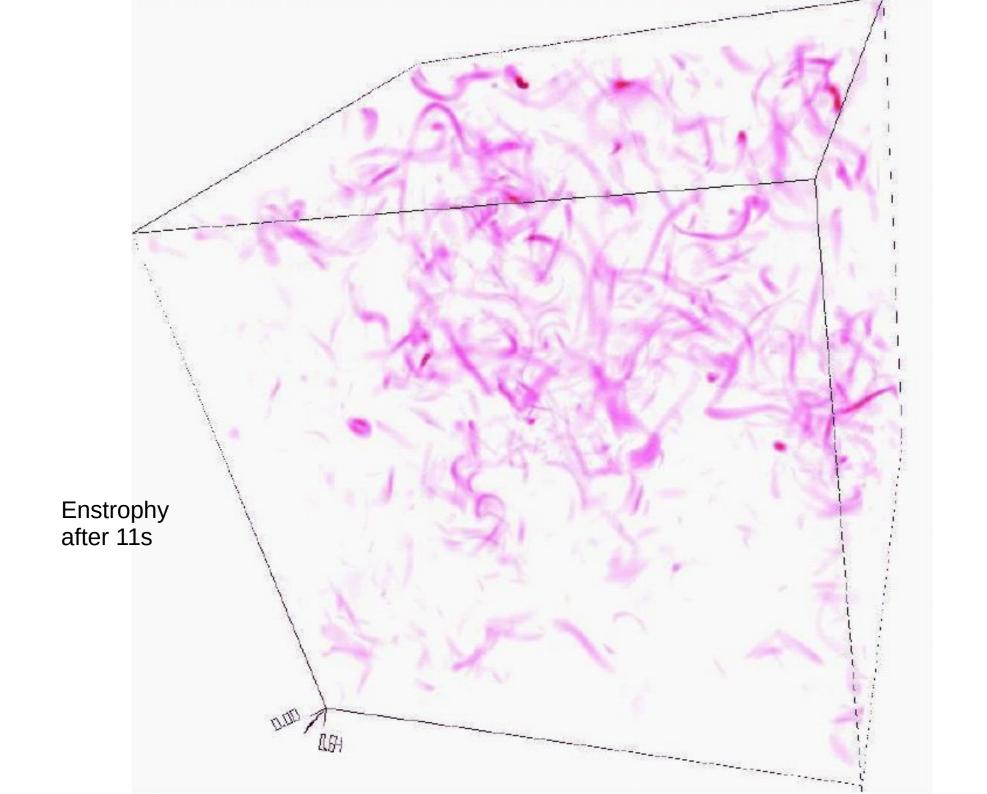
$$C_{d} = \int f \frac{dm}{dt} dr, \quad \frac{D^{*} f}{D^{*} t} = -\frac{\partial}{\partial r} \left( f \frac{dr}{dt} \right) + \eta,$$

$$D^{*} / D^{*} t \equiv \partial / \partial t + (\mathbf{v} - \mathbf{k} \mathbf{v}) \cdot \nabla$$

$$D^*/D^*t \equiv \partial/\partial t + (\mathbf{v} - \mathbf{k}v_t) \cdot \nabla$$



Cloud water after 11s



Similar equations, another application